Zn²⁺ Complexes of 3,5-Bis[(1,5,9-triazacyclododecan-3-yloxy)methyl]phenyl Conjugates of Oligonucleotides as Artificial RNases: The Effect of Oligonucleotide Conjugation on Uridine Selectivity of the Cleaving Agent

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 $2-(3,5-Bis{[1,5,9-tris(trifluoroacety])-1,5,9-triazacyclododecan-3-yloxy]methyl}phenoxy)ethanol was$ synthesized and converted to a*O*-(2-cyanoethyl)-*N*,*N*-diisopropylphosphoramidite building block,**12**. 2'-*O*-Methyl oligoribonucleotides incorporating a 2-[(2*S*,4*S*,5*R*)-4-hydroxy-5-(hydroxymethyl)tetrahydro $furan-2-yl)ethyl 4-oxopentanoate or a 2-{2-[2-({[(2$ *R*,4*S*,5*R*)-4-hydroxy-5-(hydroxymethyl)tetrahydro $furan-2-yl]acetyl}amino)ethoxy]ethoxy}ethyl 4-oxopentanoate non-nucleosidic unit close to the 3'$ terminus were assembled on a solid support, the 4-oxopentanoyl protecting groups were removed bytreatment with hydrazinium acetate on-support, and**12**was coupled to the exposed OH function. Thedeprotected conjugates were purified by HPLC, and their ability to cleave a complementary RNAcontaining either uridine or some other nucleoside at the potential cleaving site was compared.Somewhat unexpectedly, conjugation to an oligonucleotide did not enhance the catalytic activity of theZn²⁺-bis(azacrown) complex and virtually abolished its selectivity towards the uridine sites.

Introduction. – Metal ion chelates conjugated to either an oligonucleotide or a peptide nucleic acid (PNA) probe have received considerable interest as man-made restriction enzymes, with which large RNA molecules could be tailored in a sequenceselective manner [1]. Upon hybridization of the probe with its complementary target sequence, the concentration of the cleaving agent in the vicinity of one particular phosphodiester linkage is increased, resulting in accelerated chain cleavage. Since metal ion chelates are able to cleave phosphodiester bonds only within single-stranded regions [2], the cleaving agent is usually incorporated either into a terminal position of the probe, or the base sequence of the probe is designed to form upon hybridization a bulge opposite to an intrachain-cleaving agent. Owing to the proximity effect, the cleaving activity of the conjugated chelate typically is 100-fold compared to the chelate monomer [3], although, in special cases, more marked accelerations have been reported. For example, a 2'-O-methyl oligoribonucleotide bearing two 3-(3-hydroxypropyl)-1,5,9-triazacyclododecane ligands on a 2-hydroxyethyl 3'-O-(2-hydroxyethyl)- β -D-ribofuranoside branching unit (1; Fig. 1) exhibits, as a Zn²⁺ complex, 1000-fold cleaving activity compared to monomeric Zn²⁺ chelate of 1,5,9-triazacyclododecane [4], and a Cu^{2+} complex of 2,9-dimethyl-1,10-phenanthroline (2) experiences an even greater increase in catalytic activity when tethered into an intrachain position within a PNA probe creating a tetranucleotide bulge [5].

Interestingly, a base-moiety-selective cleavage may also be achieved by dinucleating ligands. (1,5,9-Triazacyclododecan-3-yl)oxy groups, when attached to an aromatic

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Fig. 1. Structures of metal ion-based cleaving agents of RNA phosphodiester linkages

scaffold, 3, cleave RNA at the 5'-side, and, to a somewhat lesser extent, at the 3'-side, of uridines [6]. Evidently, one of the Zn^{2+} -azacrown moieties anchors the cleaving agent to the uracil base, while the other azacrown moiety serves as a catalyst for the phosphodiester transesterification. Owing to this proximity effect, the cleavage is up to two orders of magnitude faster at a uridine site than at an adenosine or a cytidine site. While this difference in reaction rate is sufficient for uridine-selective cleavage of short oligonucleotides, random background cleavage still is too fast to allow controlled manipulation of sequences longer than 20-30 nucleotides [6c]. Bearing in mind that conjugation of a metal ion chelate to an oligonucleotide increases, owing to the proximity effect, its cleaving activity, one might expect this to be the case also with 3, leading to increasingly selective cleavage at a single uridine site. The present study is aimed at elucidating whether exploitation of the uracil-anchoring ability of 3 in synergy with sequence recognition by a 2'-O-methyl oligoribonucleotide probe really enhances the catalytic efficiency and, hence, ensures cleavage precisely at a single phosphodiester bond. Accordingly, oligonucleotide conjugates depicted in Fig. 2 have been prepared, and their ability to cleave various oligoribonucleotide targets has been studied.

Results. – Preparation of 2- $\{3,5-Bis[(1,5,9-triazacyclododecan-3-yloxy)methyl]$ $phenoxy}ethanol and Its Conversion to a Phosphoramidite Building Block. Commer$ cially available dimethyl 5-hydroxyisophthalate was converted to its allyl ether**4**, andthe COOMe groups were reduced to CH₂OH functions with LiAlH₄ in Et₂O(*Scheme 1*). The dimethanol**5**obtained was then tosylated in dioxane to**6**, and theTs groups were replaced by 1,5,9-tris[(*tert*-butoxy)carbonyl]-1,5,9-triazacyclododecan-3-ol (**7**), prepared as described in [6a]. The allyloxy group of the bis(azacrown)conjugate**8**was converted to 2-oxoethoxy group by OsO₄ hydroxylation andsubsequent NaIO₄ oxidation, and the resulted aldehyde**9**was reduced with NaBH₄to**10** $. Finally, the Boc protections were replaced with CF₃CO (Tfa) groups (<math>\rightarrow$ **11**), and



Fig. 2. The underlying principle of action of the artificial RNases prepared

the OH function was phosphitylated to 2-cyanoethyl *N*,*N*-diisopropylphosphoramidite **12**.

Preparation of Non-Nucleosidic Phosphoramidite Building Blocks. Non-nucleosidic building blocks allowing attachment of the bis(azacrown) conjugate as a phosphoramidite reagent, **12**, to the oligonucleotide chain were prepared as follows. 2- $[(2S,4S,5R)-5-{[Bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-hydroxytetrahydro-furan-2-yl]ethyl 4-oxopentanoate ($ **13**) was obtained as described in [7]. Its analog**17** $, with a longer side chain at C(2), was synthesized by subjecting the earlier prepared [7] ethyl [(2R,4S,5R)-5-{[bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-{[($ *tert* $-butyl)(dimethyl)silyl]oxy}tetrahydrofuran-2-yl]acetate ($ **14**) to aminolysis with 2-[2-(2-aminoethoxy)ethoxy]ethanol to obtain**15** $, acylating the OH function with (Lev)₂O (<math>\rightarrow$ **16**), and removing the TBDMS group with Bu₄NF in THF (\rightarrow **17**; Scheme 2). Compounds **13** and **17** were then converted to phosphoramidite building blocks **18** and **19**, respectively, by conventional methods.

Synthesis of Oligonucleotide Bis(azacrown) Conjugates 20 and 21. Conjugates of 2'-O-methyl oligoribonucleotides bearing the bis(azacrown) cleaving agent at the penultimate site from the 3'-terminus were assembled by the conventional phosphoramidite chemistry on a commercially available support with N⁴-benzoyl-5'-O-[bis(4methoxyphenyl)(phenyl)methyl]-2'-O-methylcytidine as the 3'-terminal nucleoside Scheme 1. Synthesis of 2-[3,5-Bis({[1,5,9-tris(trifluoroacetyl)-1,5,9-triazacyclododecan-3-yl]oxy]methyl)phenoxy]ethyl 2-Cyanoethyl N,N-Diisopropylphosphoramidite (12)



i) Allyl bromide, EtNⁱPr₂, DMF. *ii*) LiAlH₄, Et₂O. *iii*) TosCl, NaOH, H₂O, dioxane. *iv*) **7**, NaH, DMF. *v*) OsO₄, NaIO₄, H₂O, dioxane. *vi*) NaBH₄, EtOH. *vii*) 1. TfaOH, CH₂Cl₂; 2. TfaOMe, Et₃N, MeOH; 3. (Tfa)₂O, pyridine, CH₂Cl₂; 4. Et₃N, MeOH. *viii*) 2-Cyanoethyl *N*,*N*-diisopropylphosphonamidic chloride, Et₃N, CH₂Cl₂.

(Scheme 3). The 5'-O-protection was removed, and the non-nucleosidic building block, either **18** or **19**, was coupled manually using a prolonged coupling time. The Lev group was removed from the non-nucleosidic block by H_2NNH_3OAc treatment in pyridine, and the bis(azacrown) block **12** was coupled manually, using again a prolonged coupling time. The support was then loaded on the synthesizer, and the 2'-O-methyl oligoribonucleotide sequence 5'-ACA CAG ACA CGC CZC-3' (Z stands for the non-nucleosidic block) was assembled, and the support was subjected to ammonolysis to obtain the fully deprotected conjugate **20** or **21**.









Hybridization of Oligonucleotide Bis(azacrown) Conjugate **21** with Chimeric Oligoribonucleotide/2'-O-Methyl Oligoribonucleotide Targets. Table 1 records the melting temperatures for duplexes of conjugate **21** with several chimeric oligoribonucleotide/2'-O-methyl oligoribonucleotide targets, **22–27**. All the targets are fully complementary to **21** over the 3'-terminal sequence, 3'-UGU GUC UGU GCG C-5', but contain different 5'-terminal ribonucleotide overhangs. For comparative purposes, the melting temperatures of the duplexes of unconjugated 2'-O-methyl oligoribonucleotide 5'-ACA CAG ACA CGC C-3' (**28**) with the same targets are also included.

Table 1. Melting Temperatures $(T_m [^{\circ}])$ of the Duplexes of Oligonucleotide Conjugate **21** and Reference Oligonucleotide **28** with Chimeric Targets **22–27** in the Absence and Presence of Zn^{2+} Ion at pH 7.0. The data refers to [oligomer] = 2 μ M and [Zn^{2+}] = 10 μ M at the ionic strength of 0.1M, adjusted with NaCl. pH was adjusted with 10 mM phosphate buffer. Bold letters refer to ribonucleotides, normal letters to 2'-O-methyl ribonucleotides.

Target	21		28	
	with Zn ²⁺	without Zn ²⁺	with Zn ²⁺	without Zn2+
3'-UGU GUC UGU GCG GAC AAC AA-5' (22)	76.1	78.3	73.8	74.9
3'-UGU GUC UGU GCG GUA AAC AA-5' (23)	76.7	82.0	73.3	75.0
3'-UGU GUC UGU GCG GAU AAC AA-5' (24)	76.7	79.7	73.5	76.1
3'-UGU GUC UGU GCG GAA UAC AA-5' (25)	77.5	79.8	75.0	74.6
3'-UGU GUC UGU GCG GAA AUC AA-5' (26)	78.2	78.5	74.5	74.4
3'-UGU GUC UGU GCG GAA AAU AA-5' (27)	75.9	77.9	75.5	74.4

The data in *Table 1* clearly show that conjugate **21** hybridizes with all the targets even slightly better than its unconjugated counterpart **28**. The duplexes formed are stable at 35°, the temperature at which the kinetic measurements were carried out. Addition of Zn^{2+} still enhanced the hybridization of **21**, the influence being most prominent with target **23**. In this case, the melting temperature, T_m , is increased by 5.3°, while with the unconjugated reference oligonucleotide **28** the increment is 1.7°, and with target **22**, containing no uridine within the overhang, addition of Zn^{2+} increases the T_m value of the duplex with conjugate **21** by 2.2°. Evidently, the cleaving agent really recognizes the uracil base within the overhang of **23**.

Cleaving Activity of the Zn^{2+} Complexes of Oligonucleotide Bis(azacrown) Conjugates **20** and **21**. It has been shown previously [6c] that the dinuclear Zn²⁺ complex of bis(azacrown) monomer **3** cleaves oligoribonucleotide 5'-CAAUAC-3' 32 times faster than oligoribonucleotide 5'-CAACAC-3', the cleavage taking place predominantly on both sides of the uridine residue [6c]. The rate constants referring to 100 µM concentration of $3(Zn^{2+})_2$ at pH 7.5 and 35° (I=0.1M) have been reported to be $(7.70\pm0.05)\cdot10^{-6} s^{-1}$ and $(0.24\pm0.03)\cdot10^{-6} s^{-1}$, respectively. As discussed in more detail in [6], the marked rate acceleration accompanying the replacement of the intrachain cytidine with uridine may in all likelihood be attributed to anchoring of $3(Zn^{2+})_2$ through one of the azacrown chelates to the uracil base, the other chelate serving as the cleaving agent. Since oligonucleotide conjugate **21** in the presence of Zn²⁺ clearly recognizes the uracil base within the overhang of target **23**, one might expect this anchoring to lead to enhanced, highly specific cleavage of the overhang at the site of anchoring. However, this is not the case. *Table 2* contains the rate constants for the cleavage of targets 22-27 by conjugates 20 and 21 in the presence of Zn^{2+} . On the contrary, target 22 containing no uridine in the overhang was cleaved even more efficiently than the uridine-containing targets. In other words, tethering of 3 to a sequence recognizing oligonucleotide does not enhance but rather retards its uridineselective cleaving ability. Tentatively, one might speculate that the tethering of 3 results in constrain that prevents $\mathbf{3}$ to interact simultaneously with the uracil base and the adjacent phosphodiester linkage. This assumption is supported by the finding that even slightly higher cleavage rates were observed when, instead of conjugate 20 of 21, a mixture of unconjugated complementary oligonucleotide 28 and monomeric 3 was used at $[Zn^{2+}]$ of 90 µM to promote the reaction (*Table 2*). Accordingly, the situation appears to be similar to that reported in [3c] for mono(azacrown) conjugates of oligonucleotides: uracil base in the target sequence retards the cleavage by binding to the cleaving agent. Consistent with this, the slower cleavage of target 23 compared to target 24 probably results from the presence of guanosine next to uridine. It is known [6b] that $3(Zn^{2+})_2$ tends to interact with both bases in 3'-GpU-5', although a monomeric Zn^{2+} chelate binds to guanine base less firmly than to uracil base. Hence, both azacrown groups on 20 or 21 may be largely involved in anchoring to 23.

Table 2. Rate Constants ($k [10^{-6} \text{ s}^{-1}]$) for the Cleavage of Chimeric Oligonucleotides **22–27** by Complementary Oligonucleotide Conjugates **20** and **21** at pH 7.3 and 35° (I=0.1M). The data refers to [oligomer] = [**3**] = 18 µM and [Zn²⁺] = 90 µM at the ionic strength of 0.1M. pH was adjusted with a HEPES buffer. Bold letters refer to ribonucleotides, the rest to 2'-O-methyl ribonucleotides.

Target	20	21	28 + 3
3'-UGU GUC UGU GCG GAC AAC AA-5' (22)	2.44 ± 0.07	1.66 ± 0.08	0.8 ± 0.1
3'-UGU GUC UGU GCG GUA AAC AA-5' (23)	1.09 ± 0.04	0.19 ± 0.03	0.9 ± 0.1
3'-UGU GUC UGU GCG GAU AAC AA-5' (24)	2.70 ± 0.10	0.44 ± 0.03	4 ± 1
3'-UGU GUC UGU GCG GAA UAC AA-5' (25)	0.39 ± 0.04	0.25 ± 0.02	0.5 ± 0.1
3'-UGU GUC UGU GCG GAA AUC AA-5' (26)		0.82 ± 0.05	0.8 ± 0.1
3'-UGU GUC UGU GCG GAA AAU AA-5' (27)		0.79 ± 0.08	1.5 ± 0.2

Experimental Part

General. DMF and THF were dried over 3-Å, and CH_2Cl_2 , MeCN, and pyridine over 4-Å molecular sieves. Et₃N was dried by distillation and storage over CaH₂. NMR Spectra: *Bruker Avance* at 500 MHz. The chemical shifts, δ , in ppm from internal Me₄Si and the coupling constants J in Hz; appropriate 2D-NMR methods (COSY, HSQC, and HMBC) used for peak assignment. ESI-MS: in m/z.

Dimethyl 5-(Allyloxy)isophthalate (= Dimethyl 5-(Prop-2-en-1-yloxy)benzene-1,3-dicarboxylate; **4**). Dimethyl 5-hydroxyisophthalate (=dimethyl 5-hydroxybenzene-1,3-dicarboxylate; 4.9 g, 23 mmol), EtNⁱPr₂ (8.2 ml, 46 mmol), and allyl bromide (5.6 g, 46 mmol) were dissolved in DMF (5.0 ml), and the mixture was stirred at r.t. for 48 h. H₂O (50 ml) was added, and the product was extracted with Et₂O (3×30 ml). The combined org. layers were washed with sat. NaHCO₃, dried (Na₂SO₄), and evaporated to dryness. The crude product was purified by silica-gel column chromatography (CC) to yield **4** (5.0 g, 86%). Colorless solid flakes. ¹H-NMR (500 MHz, CDCl₃): 8.29 (*t*, *J* = 1.2, H–C(2)); 7.77 (*d*, *J* = 1.3, H–C(4), H–C(6)); 6.06 (*m*, CH₂CH=CH₂); 5.46 (*m*, 1 H, CH₂CH=CH₂); 5.34 (*m*, 1 H, CH₂CH=CH₂); 4.64 (*m*, CH₂CH=CH₂); 3.95 (*s*, 2 MeO). ¹³C-NMR (125 MHz, CDCl₃): 166.1 (COOMe); 158.6 (C(5)); 132.4 (CH₂CH=CH₂); 52.4 (MeO). (5-Allyloxy-1,3-phenylene)dimethanol (= [5-(Prop-2-en-1-yloxy)benzene-1,3-diyl]dimethanol; **5**). LiAlH₄ (0.93 g, 25 mmol) was slowly added to a mixture of **4** (2.1g, 8.3 mmol) in Et₂O (25 ml) at 0°. The mixture was allowed to warm up and stirred for 3 h at r.t. The reaction was quenched by addition of H₂O (100 ml), and the product was extracted with AcOEt (3×50 ml). The combined org. layers were washed with sat. aq. NaCl, dried (Na₂SO₄), and evaporated to dryness. The residue was purified by CC (5% MeOH in CH₂Cl₂) to afford **5** (1.4 g, 88%). Colorless oil. ¹H-NMR (500 MHz, CDCl₃): 6.87 (*s*, H–C(2)); 6.78 (*s*, H–C(4), H–C(6)); 6.05 (*m*, CH₂CH=CH₂); 5.41 (*m*, 1 H, CH₂CH=CH₂); 5.29 (*m*, 1 H, CH₂CH=CH₂); 15.59 (*d*, *J* = 5.0, 2 CH₂OH); 4.51 (*m*, CH₂CH=CH₂); 3.08 (*t*, *J* = 5.0, CH₂OH). ¹³C-NMR (125 MHz, CDCl₃): 158.9 (C(5)); 142.8 (C(1), C(3)); 133.2 (CH₂CH=CH₂); 117.74 (CH₂CH=CH₂); 117.71 (C(2)); 112.3 (C(4), C(6)); 68.8 (CH₂CH=CH₂); 64.8 (CH₂OH).

[(5-Allyloxy-1,3-phenylene)bis(methylene)]bis(4-methylbenzenesulfonate) (= [5-(Prop-2-en-1-yloxy)benzene-1,3-diyl]dimethanediyl Bis(4-methylbenzenesulfonate); **6**). TosCl (6.2 g, 33 mmol) in Et₂O (6.0 ml) was slowly added under stirring to a mixture of **5** (2.1 g, 11 mmol), NaOH (1.7 g, 43 mmol), H₂O (9.0 ml), and THF (5.0 ml) at 0°. Stirring was continued for 4 h at 0°, sat. aq. NaHCO₃ was added, and then the crude product was extracted with AcOEt (3×50 ml). The combined org. layers were dried (Na₂SO₄) and evaporated to dryness. CC (3% MeOH in CH₂Cl₂) of the residue yielded 3.4 g (62%) of **6** as colorless oil and 0.76 g of incompletely tosylated diol (*i.e.*, monotosylated **5** as colorless oil).

Data for **6**. ¹H-NMR (500 MHz, CDCl₃): 7.79 (m, 4 arom. H, Tos); 7.34 (m, 4 arom. H, Tos); 6.74 (m, H–C(4), H–C(6)); 6.69 (m, H–C(2)); 5.99 (m, CH₂CH=CH₂); 5.39 (m, 1 H, CH₂CH=CH₂); 5.30 (m, 1 H, CH₂CH=CH₂); 4.96 (s, 2 CH₂OH); 4.44 (m, CH₂CH=CH₂); 2.46 (s, 2 Me). ¹³C-NMR (125 MHz, CDCl₃): 158.9 (C(5)); 145.1 (C(1), C(3)); 135.3 (Tos)); 133.1 (Tos)); 132.6 (CH₂CH=CH₂); 129.9 (Tos)); 128.0 (Tos)); 120.4 (CH₂CH=CH₂); 118.0 (C(2)); 115.1 (C(4), C(6)); 71.2 (CH₂CH=CH₂); 68.9 (CH₂OH); 21.7 (Me).

Hexa(tert-butyl) 3,3'-{{[5-(2-Oxoethoxy)-1,3-phenylene]bis(methylene]}bis(oxy)}bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate) (= Hexa(tert-butyl) 3,3'-{[5-(2-Oxoethoxy)benzene-1,3-diyl]bis(methanediyloxy)}bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate); **9**). NaH (39 mg of 60% dispersion in mineral oil, 0.97 mmol) was added to a mixture of **6** (0.15 g, 0.30 mmol), tri(tert-butyl) 3-hydroxy-1,5,9-triazacyclododecane-1,5,9-tricarboxylate (**7**; 0.31 g, 0.64 mmol) and DMF. The mixture was stirred for 1 h at r.t., the reaction was quenched by slow addition of MeOH (0.5 ml) and H₂O (20 ml), and the product was extracted with Et₂O (6 × 15 ml). The org. fractions were combined, dried (Na₂SO₄), and evaporated to dryness. The residue was purified by CC (40% AcOEt in CH₂Cl₂) to yield 0.28 g (82%) of hexa(tert-butyl) 3,3'-{[[5-(allyl)oxy-1,3-phenylene]bis(methylene)]bis(oxy)}bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate) (= hexa(tert-butyl) 3,3'-{[[5-(prop-2-en-1-yloxy)benzene-1,3-diyl]bis(methanediyloxy)]bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate); **8**) as colorless oil.

Compound **8** (0.40 g, 0.35 mmol) was dissolved in a mixture of 1,4-dioxane (4.0 ml) and H₂O (1.0 ml), and a soln. of OsO₄ in 'BuOH (1:39 (*w*/*w*); 56 µl, 4.4 µmol) was added. The mixture was stirred for 30 min at r.t., cooled to 0°, a soln. of NaIO₄ (0.17 g, 0.80 mmol) in H₂O (1.0 ml) was added, and stirring was continued for 2 h. Sat. aq. NaHCO₃ (10 ml) was added to the mixture, and the product was extracted with AcOEt (3×5 ml). The combined org. layers were evaporated to dryness, and the residue was subjected to a CC (3% MeOH in CH₂Cl₂) to yield **9** (0.30 g, 74%). White foam. ¹H-NMR (500 MHz, CDCl₃): 9.85 (*s*, CH₂CH=O); 6.85 (*m*, H–C(2), H–C(4), H–C(6)); 4.62 (*s*, CH₂CH=O); 4.58 (*s*, 2 CH₂OH); 4.20–2.90 (*m*, 26 H, azacrown); 2.02 (*m*, 4 H, azacrown); 1.77 (*m*, 4 H, azacrown); 1.55 (*s*, 18 Me). ¹³C-NMR (125 MHz, CDCl₃): 199.1 (CH₂CH=O); 158.1 (C(5)); 156.3 (Boc); 156.0 (Boc); 140.5 (C(1), C(3)); 119.7 (C(2)); 112.8 (C(4), C(6)); 80.0 (CH₂CH=O); 79.7 (Boc); 75.6 (azacrown); 72.7 (CH₂O); 71.3, 53.4, 49.3, 47.0, 44.9, 29.5 (azacrown); 28.4 (Boc).

phenylene]bis(methylene)]bis(0xy)]bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate) (= hexa(tert-butyl) 3,3'-{[5-(2-hydroxyethoxy)benzene-1,3-diyl]bis(methanediyloxy)]bis(1,5,9-triazacyclododecane-1,5,9-tricarboxylate); **10**). HR-ESI-MS: 1137.7235 ($[M + H]^+$, C₅₈H₁₀₁N₆O⁺₁₆; calc. 1137.7274).

TfaOH (1.0 ml) was added to **10** (0.25 g) in CH₂Cl₂ (1.0 ml), and the mixture was stirred overnight at r.t. and evaporated to dryness. The residue was co-evaporated with H₂O, MeOH, and pyridine, dissolved in MeOH (1.0 ml), and Et₃N (0.33 ml, 2.4 mmol) and TfaOMe (0.24 ml, 2.4 mmol) were added. The mixture was stirred overnight at r.t. and evaporated to dryness. Sat. aq. NaHCO₃ (10 ml) was added, and the incompletely *N*-trifluoroacetylated products were extracted with AcOEt (3×5 ml). The combined org. layers were dried (Na₂SO₄), evaporated to dryness, and the residue was dissolved in a mixture of CH₂Cl₂ (1.0 ml) and pyridine (0.1 ml). (Tfa)₂O (0.17 ml, 1.2 mmol) was added, and the mixture was stirred overnight at r.t. and evaporated to dryness. To expose the trifluoroacetylated OH group, the residue was dissolved in MeOH (1.0 ml) containing Et₃N (50 µl), mixed for 5 h, and evaporated to dryness. NaHCO₃ (10 ml) was added to the residue, and the product was extracted with AcOEt (3×5 ml). The org. layers were combined, dried (Na₂SO₄), and evaporated to dryness. The crude product was purified by CC (40% AcOEt in CH₂Cl₂) to yield **11** (0.16 g, 54%). White foam. ¹H-NMR (500 MHz, CDCl): 6.67–6.70 (m, H–C(2), H–C(4), H–C(6)); 4.49–4.54 (m, 2 Ar-CH₂O); 3.86–4.16 (m, CH₂CH₂OH, 2 H–C(3')); 2.45–3.86 (m, 24 H, azacrown); 2.04–2.44 (m, 4 H, azacrown); 1.54–2.01 (m, 4 H, azacrown). HR-ESI-MS: 1135.2849 ([M+Na]⁺, C₄₀H₄₆F₁₈N₆NaO⁺₁₀; calc. 1135.2886.

3,3'-{{[$\{5-(2-[(2-Cyanoethoxy))(diisopropylamino)phosphinooxy]ethoxy]-1,3-phenylene}bis(methyl$ $ene)}bis(oxy)}bis(1,5,9-triazacyclododecane-1,5,9-triyl)tris(2,2,2-trifluoroethanone) (=2-[3,5-Bis({[1,5,9-tris(trifluoroacetyl)-1,5,9-triazacyclododecan-3-yl]oxy}methyl)phenoxy]ethyl 2-Cyanoethyl Dipropan-2$ ylphosphoramidoite;**12**). 2-Cyanoethyl N,N-diisopropylaminophosphoro chloridite (=3-[chloro-(diisopropylamino)phosphinooxy]propanenitrile = 2-cyanoethyl dipropan-2-ylphosphoramidochloridoite; 21 µl, 94 µmol) was added to a mixture of**11**(80 mg, 72 µmol), Et₃N (50 µl, 0.36 mmol), andCH₂Cl₂ (1.0 ml) under N₂. The mixture was stirred for 1 h at r.t. and then directly subjected to CC.Elution with Et₃N/AcOEt/petroleum ether 5:70:25 yielded**12**(80 mg, 85%). White foam. ¹H-NMR(500 MHz, CD₃CN:) 6.82 (m, H–C(2), H–C(4), H–C(6)); 4.53–4.61 (m, 2 Ar–CH₂O); 3.89–4.29 (m,OCH₂CH₂O, OCH₂CH₂CN, 2 H–C(3')); 2,97–3.89 (m, 26 H, azacrown, 2 NCHMe₂); 2.67–2.78 (m,OCH₂CH₂CN); 1.76–2.46 (m, 8 H, azacrown); 1.19–1.26 (m, 4 Me). ³¹P-NMR (200 MHz, CD₃CN):148.5. HR-ESI-MS: 1335.4007 ([M + Na]⁺, C₄₉H₆₃F₁₈N₈NaO₁₁P⁺; calc. 1335.3964).

2-[(2R,4S,5R)-5-{[Bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-{[(tert-butyl)(dimethyl)silyl]oxy]tetrahydrofuran-2-yl]-N-{2-[2-(2-hydroxyethoxy)ethoxy]ethyl]acetamide (15). Ethyl [(2R,48,5R)-5-{[Bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-{[(tert-butyl)(dimethyl)silyl]oxy}tetrahydrofuran-2-yl]acetate (14; 0.44 g, 0.71 mmol) [7] was dissolved in 1,4-dioxane (15 ml) and aq. KOH (0.20 g, 3.5 mmol in 5 ml of H₂O) was added. The mixture was stirred overnight at r.t. Solvent was removed in vacuo, the residue was dissolved in pyridine (15 ml), and the soln. was stirred with a Dowex-50 resin (pyridinium form) for 5 h. The resin was filtered off, pyridine was evaporated, and, after co-evaporation with dry pyridine, the residue was dissolved in dry DMF (2 ml). EtNⁱPr₂ (0.15 ml, 0.85 mmol), HBTU (0.32 g, 0.85 mmol), and 2-[2-(2-aminoethoxy)ethoxy]ethanol (0.15 g, 0.85 mmol) were added, and the mixture was strirred for 2.5 h at r.t. DMF was removed by evaporation, the residue was dissolved in Et₂O, washed with sat. NaHCO₃, dried (Na₂SO₄), and evaporated to dryness. Purification by CC (0.1% Et₃N and 3% MeOH in CH₂Cl₂) yielded 15 (0.33 g, 65%). Yellowish oil. ¹H-NMR (500 MHz, CDCl₃): 7.23 -7.46 (m, 9 H, DMTr); 6.85 (d, J = 8.8, 4 H, DMTr); 6.78 (t, J = 5.5, NH); 4.46 - 4.49 (m, H–C(5)); 4.25 -4.26 (m, H-C(2)); 3.95-3.97 (m, H-C(4)); 3.81 (s, 6 H, DMTr); 3.42-3.69 (m, NHCH₂CH₂(OCH₂. CH₂)₂OH); 3.14 (dd, J = 9.9, 4.5, 1 H, C(5)–CH₂O); 3.09 (dd, J = 9.9, 5.3, 1 H, C(5)–CH₂O); 2.53 (dd, J = 15.1, 3.8, 1 H, C(2)–CH₂C=O); 2.46 (dd, J = 15.1, 8.1, 1 H, C(2)–CH₂C=O); 1.91 (ddd, J = 12.8, 5.2, 1.2) 1.3, 1 H, CH₂(3)); 1.71 (m, 1 H, CH₂(3)); 0.86 (s, 'Bu); 0.03 (s, MeSi); 0.01 (s, MeSi). ¹³C-NMR (125 MHz, CDCl₃): 171.0 (C=O); 158.5, 144.8, 136.1, 130.1, 128.2, 127.8, 126.8, 113.1, 86.9 (DMTr); 86.1 (C(5)); 75.1 (C(4)); 74.1 (C(2)); 72.5, 70.3, 70.2, 69.9 (NHCH₂CH₂OCH₂CH₂OCH₂CH₂OCH₂CH₂OH); 64.1 $(C(5)-CH_2O);$ 61.7 $(OCH_2CH_2OH);$ 55.2 (DMTr); 42.5 (C(2)); 41.1 $(C(2)-CH_2C=O);$ 39.1 (NCH₂CH₂O); 25.8 (TBDMS); 18.0 (TBDMS); -4.7 (TBDMS); -4.8 (TBDMS). HR-ESI-MS: 746.3642 ($[M + Na]^+$, $C_{40}H_{57}NNaO_9Si^+$; calc. 746.3695).

2-{2-[2-({[(2R,4S,5R)-5-{[Bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-{[(tert-butyl)(dimethyl)silyl]oxy}tetrahydrofuran-2-yl]acetyl]amino)ethoxy]ethoxy]ethyl 4-Oxopentanoate (16). Levulinic anhydride ((Lev)₂O; 0.29 g, 1.4 mmol) was added to a soln. of **15** (0.33 g, 0.46 mmol) in pyridine. The mixture was stirred overnight at r.t. and evaporated to dryness. The residue was dissolved in CH₂Cl₂ and washed with sat. aq. NaHCO₃. The org. phase was dried (Na₂SO₄) and evaporated to dryness. The crude product was purified by CC (0.1% Et₃N and 3% MeOH in CH₂Cl₂) to yield 16 (0.32 g, 85%). ¹H-NMR (500 MHz, CDCl₃): 7.21 – 7.45 (*m*, 9 H, DMTr); 6.84 (*d*, *J* = 8.9, 4 H, DMTr); 6.69 (*t*, *J* = 5.5, NH); 4.43 – 4.50 (m, H-C(5)); 4.25 - 4.26 (m, H-C(2)); 4.21 (t, J = 4.8, OCH₂CH₂OLev); 3.94 - 3.96 (m, H-C(4));3.81 (s, 2 MeO); 3.43 - 3.64 (m, NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev); 3.12 (dd, J = 9.9, 4.5, 1 H, $C(5)-CH_2O$; 3.07 (dd, J = 9.9, 5.3, 1 H, $C(5)-CH_2O$); 2.75 (t, J = 6.5, 2 H, Lev); 2.60 (t, J = 6.5, 2 H, Lev); 2.52 $(dd, J = 15.1, 4.0, 1 \text{ H}, C(2) - CH_2C = O)$; 2.45 $(dd, J = 15.1, 8.1, 1 \text{ H}, C(2) - CH_2C = O)$; 2.20 (s, d)3 H, Lev); 1.89-1.93 (m, 1 H, CH₂(3)); 1.67-1.73 (m, 1 H, CH₂(3)); 0.86 (s, 'Bu); 0.02 (s, MeSi); 0.01 (s, MeSi). ¹³C-NMR (125 MHz, CDCl₃): 206.6 (Lev); 172.7 (Lev); 170.9 (CH₂C(O)NH); 158.5, 144.8, 136.0, 130.1, 128.2, 127.8, 126.8, 113.1, 86.9 (DMTr); 86.0 (C(5)); 75.1 (C(4)); 74.1 (C(2)); 70.5, 70.2, 69.9, 69.0 (NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev); 64.1 (OCH₂CH₂OLev); 63.7 (C(5)-CH₂O); 55.2 (DMTr); 42.6 (C(3)); 41.3 (C(2)-CH₂C=O); 39.2 (NHCH₂CH₂O); 37.9 (Lev); 29.9 (Lev); 27.9 (Lev); 25.8 (TBDMS); 18.0 (TBDMS); -4.6 (TBDMS); -4.8 (TBDMS). HR-ESI-MS: 844.4068 ([M+Na]+, C₄₅H₆₃NNaO₁₁Si⁺; calc. 844.4063).

an-2-yl]acetyl]amino)ethoxy]ethoxy]ethyl 4-Oxopentanoate (17). $Bu_4NF \cdot H_2O$ (0.30 g, 1.1 mmol) was added to a soln. of 16 (0.31 g, 0.38 mmol) in THF (3 ml). The mixture was stirred for 2 h at r.t., and then volatiles were removed. The residue was dissolved in CH2Cl2 and washed with H2O. The org. phase was dried (Na₂SO₄) and evaporated to dryness. The crude product was purified by CC (0.1% Et₃N and 3% MeOH in CH₂Cl₂) to yield **17** (0.21 g, 78%). ¹H-NMR (500 MHz, CDCl₃): 7.23 – 7.46 (*m*, 9 H, DMTr); 6.84 (d, J = 8.9, 4 H, DMTr); 6.68 (t, J = 5.5, NH); 4.46 - 4.53 (m, H-C(5)); 4.33 - 4.34 (m, H-C(2)); 4.21 $(t, J = 4.8, \text{ OCH}_2\text{CH}_2\text{OLev}); 3.98 - 4.00 (m, H-C(4)); 3.81 (s, 6 H, DMTr); 3.44 - 3.65 (m, H)$ NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev); 3.24 (dd, J = 9.7, 4.7, 1 H, C(5)–CH₂O); 3.11 (dd, J = 9.7, 5.7, 1 1 H, C(5)– CH_2O); 2.75 (t, J = 6.5, 2 H, Lev); 2.60 (t, J = 6.5, 2 H, Lev); 2.41–2.54 (m, C(2)– $CH_2C=O$); 2.19 (s, 3 H, Lev); 2.04 (ddd, J = 15.1, 5.6, 2.0, 1 H, CH₂(3)); 1.81-1.87 (m, 1 H, CH₂(3)). ¹³C-NMR (125 MHz, CDCl₃): 206.8 (Lev); 172.8 (Lev); 170.7 (CH₂C(O)NH); 158.5, 144.8, 136.0, 130.1, 128.2, 127.8, 126.8, 113.1, 86.2 (DMTr); 86.2 (C(5)); 75.0 (C(4)); 74.2 (C(2)); 70.5, 70.3, 69.8, 69.0 (NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev); 64.5 (OCH₂CH₂OLev); 63.7 (C(5)–CH₂O); 55.2 (DMTr); 42.6 (C(3)); 40.8 (C(2)-CH₂C=O); 39.2 (NHCH₂CH₂O); 37.9 (Lev); 29.9 (Lev); 27.9 (Lev). HR-ESI-MS: 730.3355 ($[M + Na]^+$, $C_{39}H_{49}NNaO_{11}^+$; calc. 730.3198).

2-{2-[2-({[(2R,4S,5R)-5-{[Bis(4-methoxyphenyl)(phenyl)methoxy]methyl}-4-({(2-cyanoethoxy)[di-(propan-2-yl)amino]phosphanyl]oxy)tetrahydrofuran-2-yl]acetyl]amino)ethoxy]ethoxy]ethyl 4-Oxopentanoate (19). 2-Cyanoethyl N,N-diisopropylaminophosphorochloridite (30 µl, 0.14 mmol) was added to a mixture of 17 (74 mg, 0.10 mmol) and Et_3N (73 μ l, 0.52 mmol) in CH_2Cl_2 (0.5 ml) under N₂. The mixture was stirred for 1 h at r.t. and then subjected directly to CC (5% Et₃N in AcOEt) to give 19 (48 mg, 51%). Colorless oil. ¹H-NMR (500 MHz, CD₃CN): 7.23 – 7.48 (*m*, 9 H, DMTr); 6.86 – 6.89 (*m*, 4 H, DMTr); 6.62 $(t, J=4.4, \text{ NH}); 4.37-4.41 (m, H-C(2), H-C(5)); 4.12 (t, J=4.7, OCH_2CH_2OLev); 3.95-4.02 (m, H-C(2)); 4.12 (t, J=4.7, OCH_2CH_2OLev); 4.12 (t, J=4.7, OCH_2OLev); 4.12 (t, J=4.7, OCH_2OLev); 4.12 (t, J=4.7, OCH_2OLev$ H-C(4)); 3.77 (s, 6 H, DMTr); 3.28-3.76 (m, NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev, OCH₂CH₂CN); $3.10-3.16 (m, 1 H, C(5)-CH_2O); 3.02-3.05 (m, 1 H, C(5)-CH_2O); 2.70 (t, J = 6.4, 2 H, Lev); 2.62 (m, 1 H, C(5)-CH_2O); 2.62 (m$ 2 NCHMe_{2} ; 2.52 (t, J = 6.4, 2 H, Lev); 2.48 (m, $OCH_{2}CH_{2}CN$); 2.42–2.40 (m, 1 H, $C(2)-CH_{2}C=0$); 2.15-2.04 (m, 1 H, C(2)-CH₂C=O); 2.10 (s, 3 H, Lev); 1.98-1.94 (m, 1 H, CH₂(3)); 1.78-1.85 (m, 1 H, CH₂(3)); 1.05-1.25 (m, 2 NCHMe₂). ¹³C-NMR (125 MHz, CD₃CN) 206.8 (Lev); 172.6 (Lev); 170.1 (CH₂C(O)NH); 158.7, 145.3, 136.1, 130.1, 128.1, 127.8, 126.8, 113.1, 85.9 (DMTr); 85.5 (C(5)); 75.6 (C(4)); 75.1 (C(2)); 70.2, 69.9, 69.3, 68.7 (NHCH₂CH₂OCH₂CH₂OCH₂CH₂OLev); 64.2 (OCH₂CH₂OLev); 63.5 $(C(5)-CH_2O); 60.0 (OCH_2CH_2CN); 54.9 (MeO); 43.0 (NCHMe_2); 42.9 (C(3)); 41.9 (C(2)-CH_2C=O);$ 39.6 (NHCH₂CH₂O); 38.8 (Lev); 29.0 (Lev); 27.7 (Lev); 23.9 (CHMe₂); 20.1 (OCH₂CH₂CN). ³¹P-NMR $(200 \text{ MHz}, \text{CD}_3\text{CN}) 147.3, 147.2, \text{HR-ESI-MS}: 930.4545 ([M + \text{Na}]^+, \text{C}_{48}\text{H}_{66}\text{N}_3\text{NaO}_{12}\text{P}^+; \text{calc}, 930.4276).$

Synthesis of Oligonucleotide Conjugates 20 and 21. Oligonucleotide conjugates 20 and 21 were assembled by the phosphoramidite chemistry on a commercially available support bearing DMTr-

protected N⁴-benzoyl-2'-O-methylcytidine, 8 (1.0 µmol) as the 3'-terminal nucleoside. The DMTr group was manually removed with 3% Cl₂CHCOOH in CH₂Cl₂, and the deprotected support was washed with CH₂Cl₂ and MeCN and dried. The non-nucleosidic building block, either 18 or 19 (10 µmol in 100 µl of MeCN) was coupled manually using 5-(benzylthio)tetrazole (50 µl, 0.25M in MeCN) as activator and 0.5 h coupling time. Another portion of activator was added, and the reaction was allowed to proceed for another 0.5 h. The support was washed with MeCN, CH₂Cl₂, and THF, and then subjected to the capping and oxidation steps according to the standard RNA coupling protocol. The Lev protecting groups of the support-bound non-nucleosidic block were removed with 0.5 m H₂NNH₃OAc in pyridine (NH₂NH₂·H₂O/ pyridine/AcOH 0.124:4:1, 0.5 h). The support was washed with pyridine and MeCN, and dried. The azacrown building block 12 (11 mg, 8.6 µmol in 33 µl of MeCN) was then coupled manually using 5-(benzylthio)tetrazole (50 μl, 0.25M in MeCN) as activator. The mixture was kept 2 h at r.t., and then another portion of activator was added. After 1 h, the support was washed with MeCN, CH₂Cl₂, and THF, and subjected to the capping and oxidation steps according to the standard RNA coupling protocol. The 2'-O-methyl oligonucleotide sequence was then assembled on an Applied Biosystems 3400 DNA synthesizer in 1.0-µmol scale applying the conventional phosphoramidite chemistry and the standard RNA coupling protocol. The fully protected 2'-O-methyl oligoribonucleotide conjugates 20 and 21 were released from the support and deprotected with conc. aq. NH₃ (5 h at 55°). The crude product mixture was filtered, and the filtrate was evaporated to dryness, dissolved in H₂O, and then subjected to RP-HPLC purification. The authenticity of the conjugate was verified by ESI-MS spectroscopy. 20: ESI-HR-MS: 1808.0987 ($[M - 3 H]^{3-}$), giving 5426.31 ($[M - H]^{-}$; calc. 5426.30). **21**: ESI-HR-MS: 1857.2 ($[M - H]^{-}$) $([M-4H]^{4-}), 1113.9 ([M-5H]^{5-}), giving 5573.5 ([M-H]^{-}; calc. 5573.4).$

Chimeric 2'-O-methyl oligoribonulceotides/oligoribonucleotides 22-27, used as targets for the artificial ribonucleases 20 and 21, and 2'-O-methyl oligoribonucleotide 28, used as a reference material for the melting-temperature (T_m) studies, were assembled from commercially available 2'-O-methyl and 2'-O-[(triisopropylsilyloxy)methyl]-protected 2-cyanoethyl *N*,*N*-diisopropylphosphoramidite building blocks (*Glen Research*) by the conventional phosphoramidite strategy using a 1.0-µmol scale and applying the standard RNA coupling protocol of *Applied Biosystems 392* or 3400 DNA synthesizer. All buffer solns. were prepared in sterilized H₂O, and sterilized equipment was used for their handling.

 T_m Studies. The melting curves (absorbance vs. temp.) were recorded at 260 nm on a Perkin-Elmer Lambda 35 UV/VIS spectrometer equipped with a multiple cell holder and a Peltier temp. controller. The temp. was changed at a rate of 0.5°/min (from 15 to 90°). The recordings were performed in 10 mm potassium phosphate buffer (pH 7) containing 0.1M NaCl. The oligonucleotide conjugates **20** and **21**, reference oligonucleotide **28**, and their targets **22–27** were used at a concentration of 2 μ M. The T_m values were determined as the maximum of the first derivative of the melting curve.

Kinetic Measurements. The reactions were carried out in Eppendorf tubes immersed in a water bath, the temp. of which was kept at $35.0 \pm 0.1^{\circ}$. The pH was adjusted to 7.3 with a *HEPES* buffer (0.1M), and the Zn^{2+} ion was added as a nitrate salt to give the total metal ion concentration of 90 μ M. The ionic strength was adjusted to 0.1m with NaNO3. The concentrations of the azacrown-functionalized oligonucleotides and their targets were 18 µM. 4-Nitrobenzenesulfonate ion was used as an internal standard. The total volume of the reaction mixture was 200 µl in each kinetic run. Aliquots of 20 µl were withdrawn at suitable intervals, and the reaction was quenched by adding aq. HCl (1.0 µl of 1.0M soln.). The samples were analyzed immediately by cap. zone electrophoresis (Beckman Coulter P/ACE MDQ CE System) using a fused silica capillary (inner diameter 50 µm, effective length 50 cm). The inverted polarity, citrate buffer (0.2м, pH 3.1), and – 30 kV voltage were used. The temp. of the capillary was kept at 25°. The samples were injected using hydrodynamic injection with 2 psi for 8 s. The capillary was flushed for 3 min with H_2O , 10 mM aq. HCl, and the background electrolyte buffer between every anal. run. The quantification of the target and product oligonucleotides was based on comparison of their UV absorption at 254 nm to that of the internal standard. The peak area was first normalized by dividing it by the migration time and then by the similarly normalized peak area of the internal standard. First-order rate constants for the cleavage of the target oligonucleotides were calculated by applying the integrated first-order rate law to the disappearance of the starting material.

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Received March 26, 2012